

LITERATURE CITED

1. A. L. Markman and V. P. Rzhekhin, *Gossypol and Its Derivatives* [in Russian], Moscow (1965), p. 23.
2. R. T. Grigorchuk, E. B. Skomorokhova, A. N. Mironova, and U. I. Tros'ko, *Maslob-Zhir. Promst.*, No. 11, 11 (1979).
3. Kh. K. Kholmatov, T. V. Chernenko, A. I. Glushenkova, and A. U. Umarov, *Abstracts of Lectures at a Conference on Problems of Extraction from Solids* [in Russian], Tashkent (1977).

COMPONENTS OF *Artemisia leucodes*

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We have continued the study of the chemical composition of the epigeal part of *Artemisia leucodes* Schrenk collected in the budding stage in the Chemkent province, Kazakh SSR [1].

The chromatographic separation of the total lactone material on neutral alumina followed by rechromatography of the individual fractions in a silica gel column yielded two substances.

Substance I with mp 176-178°C was readily soluble in all organic solvents. A comparison of IR spectra and a mixed melting point showed that compound (I) was β -camphor, which has been isolated from the essential oil of *A. leucodes* previously [2].

Substance (II) had mp 190-191°C (from ethanol) and the composition $C_{17}H_{26}O_5$, with R_f 0.43 on Silufol in the chloroform-ethyl acetate (9:1) system. The IR spectrum of (II) contained absorption bands at (cm^{-1}) 1790 (carbonyl of a γ -lactone ring), 1745, 1240 (carbonyl of an ester group), 1685 (α,β -unsaturated cyclopentanone), and 1640 and 1625 (conjugated double bond). The facts given for (II) correspond to the sesquiterpene lactone matricarin [3]. A direct comparison of (II) with matricarin which we obtained by the acetylation of austricin, showed their identity.

This is the first time that matricarin has been isolated from *Artemisia leucodes*.

From *A. leucodes* in the flowering stage, together with sesquiterpene lactones [4], we isolated yellow needles with mp 227-228°C (benzene-chloroform), having the composition $C_{18}H_{16}O_7$, which were assigned, on the basis of their UV spectrum (λ_{max} 252, 277, 345 nm) and qualitative reactions to the flavone derivatives. It was established from the PMR spectrum that the flavonoid contained three OCH_3 groups (3.66, 3.70, and 3.82 ppm, 3 H each). The presence in the mass spectrum of the peaks of ions with m/z 165 and 162 showed that two of the OCH_3 groups are present in ring B [5].

Analysis of the PMR, mass, and UV spectra, and also a comparison of the physicochemical properties with literature figures enabled the flavonoid isolated to be identified as eupatilin [5, 6]. Eupatilin has been isolated previously from *Artemisia frigida* [7].

LITERATURE CITED

1. I. M. Saitbaeva, A. Mallabaev, and G. P. Sidyakin, *Khim. Prir. Soedin.*, 56 (1983).
2. M. I. Goryaev, V. S. Bazalitskaya, and P. P. Polyakov, *Chemical Composition of Wormwoods* [in Russian], Alma-Ata (1962), p. 94.
3. W. Herr and K. Ueda, *J. Am. Chem. Soc.*, **83**, 1139 (1961).
4. I. M. Saitbaeva, A. Mallabaev, and G. P. Sidyakin, *Khim. Prir. Soedin.*, 247 (1981).
5. M. Goudard, J. Favre-Bonvin, P. Lebreton, and J. Chopin, *Phytochemistry*, 145 (1978).
6. M. Kupchan, C. W. Sigel, et al., *Tetrahedron*, 1603 (1969).
7. G.-L. Lin and T. J. Mabry, *Phytochemistry*, 1389 (1981).

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